

中國文化大學 101 學年度碩博士班甄試入學招生考試

系組：化學系應用化學碩士班日期 100 年 11 月 23 日 節次：第 1 節

科目：綜合化學

1. A one-dimensional box is located from $x=0$ to $x=L$. The potential energies at the boundaries are ∞ , but 0 inside the box. Now a particle is confined in the box and its wavefunction is $\Psi = \sqrt{\frac{2}{L}} \sin \frac{n\pi x}{L}$, why should n be nonzero integers? (10%)
2. Acetic acid has $K_a = 1.8 \times 10^{-5}$, now mix 500 ml of 0.1M acetic acid and 500 ml of 0.1M sodium acetate to prepare a buffer solution, what will be the change of pH after the buffer is added with 100 ml of 0.1M HCl? (10%)
3. With the VSEPR model, please predict the structure of XeF_4 . (10%)
4. An aqueous solution gives rise to an osmotic pressure of 4.00 atm at 27°C , what is the molarity of that solution? (10%)
5. Complete the following reactions : (5% for each)
 - (a) $\text{CH}_3\text{CONHCH}_3 + \text{H}_2\text{O} \xrightarrow{\text{H}^+} ?$
 - (b) $\text{C}_6\text{H}_6 + \text{SO}_3 \xrightarrow{\text{H}_2\text{SO}_4} ?$
6. Is O_2 paramagnetic or diamagnetic? why? (10%)
7. The rate constants for $2\text{N}_2\text{O}_5(\text{g}) \rightarrow 4\text{NO}_2(\text{g}) + \text{O}_2(\text{g})$ are $3.7 \times 10^{-5} \text{ s}^{-1}$ at 25°C and $1.7 \times 10^{-3} \text{ s}^{-1}$ at 55°C . What is the activation energy in kJ/mol? (10%)
8. Why do cylindrical carbon nanotubes really exist while cylindrical silicon nanotubes may not? (10%)
9. F^- is a weak field ligand and CN^- is a strong field ligand, why is CoF_6^{3-} high spin and $\text{Co}(\text{CN})_6^{3-}$ low spin according to the crystal field theory? (10%)
10. What is transcription of DNA? What is translation of RNA? (5% for each)