

I. Please describe the usefulness of the following materials or techniques in biochemical experiments. (20%)

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|-----------------------------|----------------------------------|
| 1. Sodium dodecyl sulfate   | 6. Southern blotting             |
| 2. DEAE-Cellulose           | 7. Size exclusion chromatography |
| 3. EDTA                     | 8. Ninhydrin                     |
| 4. Ethidium bromide         | 9. Edman degradation             |
| 5. Isopycnic centrifugation | 10. Dideoxynucleotide            |

II. What is the main biochemical or physiological function of the following compounds. (20%)

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|------------------------------|---------------------------------|
| 1. Adenosine-5'-triphosphate | 6. Biotin                       |
| 2. pyridoxal-5'-phosphate    | 7. Adenylate cyclase            |
| 3. Immunoglobin G            | 8. Aminoacyl-tRNA synthase      |
| 4. Coenzyme A                | 9. Ribosome                     |
| 5. Reverse transcriptase     | 10. Flavin adenine dinucleotide |

III. Please define the following terms: (20%)

- |                              |                                  |
|------------------------------|----------------------------------|
| 1. Turnover number           | 6. Beer's law                    |
| 2. Avogadro's number         | 7. Chargaff's rule               |
| 3. Michaelis constant        | 8. Svedberg unit                 |
| 4. Sedimentation coefficient | 9. Lineweaver-Burk plot          |
| 5. Hyperchromic effect       | 10. Molar absorption coefficient |

IV. The density of water at 4°C is 1.0. Please calculate the concentration (in molarity) of water under this condition. (5%)

V. The hydrochloric acid is a strong electrolyte. It exists almost in the ionic state in a dilute solution. What is the pH value of 1mM HCl in water? (5%)

VI. The equilibrium constant ( $K_{eq}$ ) of transamination reaction is about 1.0 at 25°C. Please calculate the  $\Delta G^\circ$  values under this condition. (10%)

VII. Protein is made of 20 amino acids. Please give the name and the structure of the amino acid which contains no asymmetric carbon. (10%)

VIII. How do you define the pI? The followings are  $pK_a$ 's of L-lysine:  $pK_a(\alpha\text{-COOH}) = 2.2$ ,  $pK_a(\alpha\text{-NH}_3^+) = 9.2$  and  $pK_a(\epsilon\text{-NHL}_2^-) = 10.8$ . Please estimate its pI value. (10%)

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