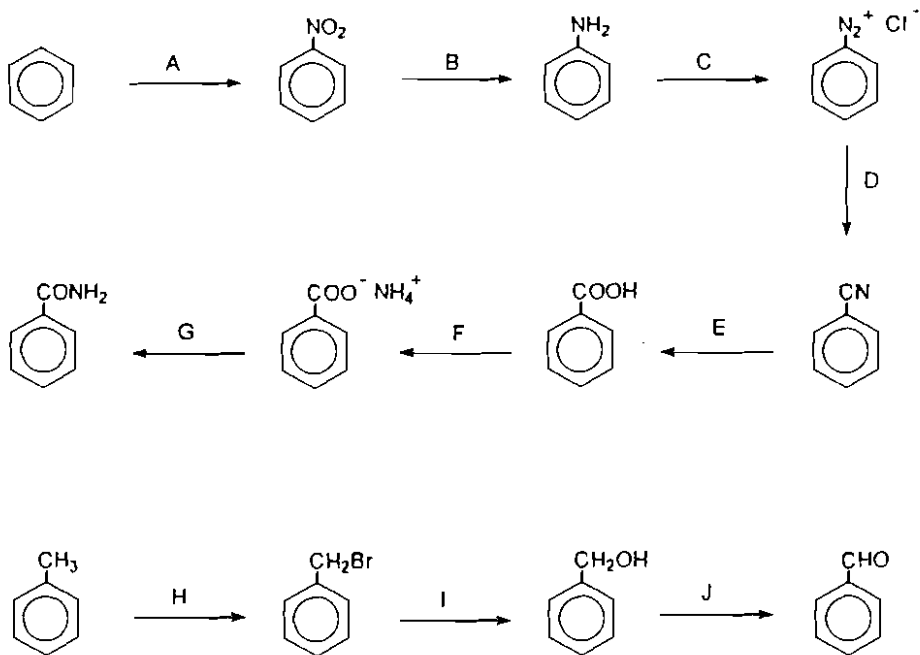
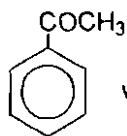


一、 Supply the suitable reagents for the following conversion. (2 pts. each; 20 pts. total)

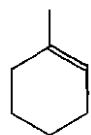


二、 Reactions. (3 pts. each; 30 pts. total)



with

- A. $I_2, NaOH \longrightarrow$
 B. $NaCN, acid \longrightarrow$
 C. excess $CH_3OH, acid \longrightarrow$
 D. $NH_2NH_2, KOH, heat \longrightarrow$



with

- E. $OsO_4 \longrightarrow$
 F. $H_2SO_4 \longrightarrow$
 G. $Br_2/H_2O \longrightarrow$
 H. $HBr, peroxide \longrightarrow$
 I. $KMnO_4, base, heat \longrightarrow$

三、 Explain/Define the following terms:(2 pts each; 10 pts total)

- (1) Beer's Law (2) Blind sample (3) Absorption (4) HETP
(5) Reversed-phase chromatography

四、 What is the purpose of a matrix modifier in atomic spectroscopy? (10 pts)

五、 The molar absorptivities of X and Y were measured with pure samples of each:
(5 pts each; 10 pts total)

λ (nm)	ϵ ($M^{-1} \text{ cm}^{-1}$)	
	X	Y
$\lambda^1=406$	$\epsilon x^1=720$	$\epsilon y^1=212$
$\lambda^2=457$	$\epsilon x^2=479$	$\epsilon y^2=274$

A mixture of X and Y in a 1.0 cm cell had an absorbance of $A^1=0.722$ at 406 nm and $A^2=0.641$ at 457 nm. Find the concentrations of X and Y in the mixture.

六、 How do you confirm the accuracy of the analysis method? (10 pts)

七、 Briefly compare the LLE, SPE and SPME for the extraction liquid sample. (10